

BIOSYNTHESIS OF EPHEDRINE IN EPHEDRA

PARTICIPATION OF C₆-C₁ UNIT

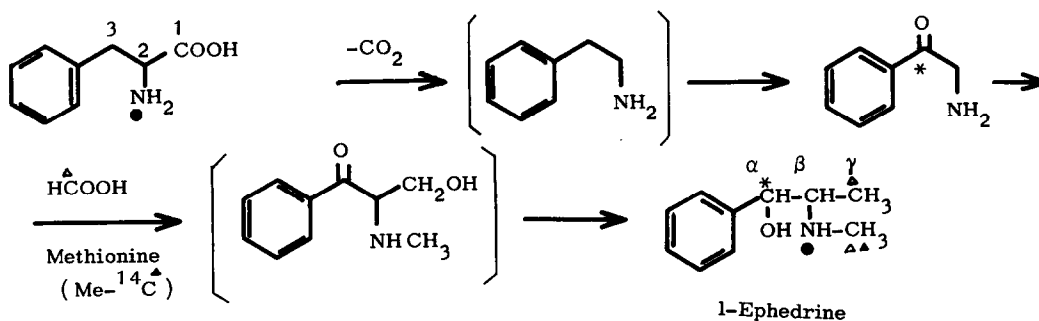
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In the previous studies on the biosynthesis of ephedrine in *Ephedra distachya* L. (Ephedraceae) Shibata et al.¹⁻⁴ described that ¹⁵N of labelled phenylalanine was incorporated into nitrogen of l-ephedrine and ¹⁴C-methyl group of methionine into N-methyl group, while ¹⁴C of formate was distributed 42% in N-methyl and 37% in C-methyl (at the γ-position of side chain), and carbonyl-¹⁴C of ω-aminoacetophenone was highly incorporated into the α-position of ephedrine side chain.

On the basis of these results, the following scheme was proposed for the biosynthesis of l-ephedrine in Ephedra plant.



However, our renewed experiments revealed that phenylalanine labelled at 2-¹⁴C was not incorporated into l-ephedrine, whereas aromatic ³H and 3-¹⁴C were introduced in it.

The location of radioactivity was proved by the degradation of l-ephedrine by Kuhn-Roth oxidation into benzoic acid and acetic acid. The latter was characterized as the p-bromophenacyl ester.

Table I
Incorporation of DL-Phenylalanine into l-Ephedrine

Labelled position	Incorporation ratio	Specific activities of degradation products (100 = Specific activity of l-ephedrine)	
		Benzoic acid (ϕ -C $_{\alpha}$)	Acetic acid (C $_{\beta}$ -C $_{\gamma}$)
Aromatic- ^3H	2.9×10^{-4}	---	*
2- ^{14}C	nil	---	---
3- ^{14}C	1.4×10^{-4}	---	---
{ Aromatic- ^3H 2- ^{14}C	5.3×10^{-4}	106	---
	4.0×10^{-5}	---	---
{ Aromatic- ^3H 3- ^{14}C	8.5×10^{-4}	---	---
	8.4×10^{-4}	100	nil
{ 2- ^{14}C 3- ^{14}C	8.2×10^{-5}	92	nil

* --- not measured

These results suggest that phenylalanine fed to the plant is cleaved between C₍₂₎ and C₍₃₎, and only C₆-C₁ portion is used for the biosynthesis of l-ephedrine. Then sodium benzoate (Carboxyl- ^{14}C), benzaldehyde(Carbonyl- ^{14}C) and sodium cinnamate-(3- ^{14}C) were administrated to Ephedra. All these labelled compounds were incorporated in higher ratios than phenylalanine into the corresponding positions of ephedrine molecule.

Table II
Incorporations of Benzoate, Benzaldehyde and Cinnamate into l-Ephedrine

Precursors	Incorporation ratio	Specific activities of degradation products (100 = Specific activity of l-ephedrine)	
		Benzoic acid (ϕ -C $_{\alpha}$)	Acetic acid (C $_{\beta}$ -C $_{\gamma}$)
Benzoate (Carboxyl- ^{14}C)	1.3×10^{-1}	104	nil
Benzaldehyde(Carbonyl- ^{14}C)	1.6×10^{-2}	96	nil
Cinnamate(3- ^{14}C)	1.2×10^{-3}	100	nil

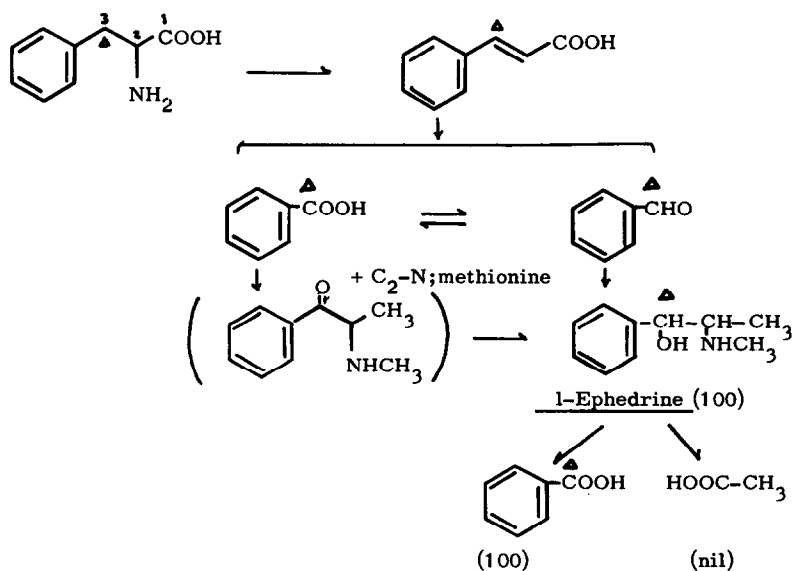
Mescaline in Cactus, hordenine in germinating barley and epinephrine in the medulla of the adrenals were experimentally proved to be biosynthesized from tyrosine via C₆-C₂-N intermediates. In contrast with them, ephedrine is biosynthesized from C₆-C₁ and C₂-N(+C₁) units.

The existence of benzylmethylamine was reported⁵⁾ in *Ephedra*, which would suggest the possibility of occurrence of C₆-C₁ intermediate.

The incorporation of phenylalanine-(3¹⁴C) into d-norpseudoephedrine in *Catha edulis* reported by Leete⁶⁾ is not incompatible with our new scheme of ephedrine biosynthesis.

Ephedrine and its homologues contained in *Ephedra* plants have S-configuration about the carbon atom at 3-position of the side chain. Contrary to this, chloramphenicol has R-configuration in respect to the corresponding position and it is biosynthesized from L-p-aminophenylalanine without decarboxylation.⁹⁾ If L-serine or L-alanine are combined with benzoic acid or benzaldehyde, the configuration of the β-carbon atom must be S. A remarkable distribution (37%) of the radioactivity of formate-¹⁴C into the γ-carbon of ephedrine, which was shown in the previous experiment³⁾ could be explained by the contribution of formate via serine. The feeding experiment using L-serine(U-¹⁴C) and L-alanine(U-¹⁴C) showed poor incorporation into ephedrine (5.4 x 10⁻³, and 7.4 x 10⁻⁴%, respectively) and randomization of radioactivity giving no conclusive evidence for the origin of C₂-N unit. However, this cannot exclude the above possibility, since similar example of the formate incorporation was observed at the tryptophan moiety of evodiamine, a Rutaceous alkaloid⁷⁾.

The incorporation of ω-aminoacetophenone(Carbonyl-¹⁴C) into ephedrine can be explained by its unstable nature decomposing into C₆-C₁ unit, and the incorporation of ¹⁵N-labelled phenylalanine would be resulted by transamination. By the present experimental results the scheme of ephedrine biosynthesis proposed previously has to be amended, and the following pathway is now presented:



A biogenetical hypothesis involving the condensation of benzaldehyde and N-methylalanine was suggested by Akabori and Momotani⁸⁾ on the basis of their chemical synthesis of dl-pseudo-ephedrine.

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References

1. S. Shibata and I. Imaseki : Chem.Pharm.Bull.(Tokyo), 4 , 277 (1956)
2. S. Shibata, I. Imaseki and M. Yamazaki: Chem.Pharm.Bull.(Tokyo), 5 , 71 (1957)
3. S. Shibata, I. Imaseki and M. Yamazaki: Chem.Pharm.Bull.(Tokyo), 5 , 594 (1957) ; Chem. & Ind., 1958 , 1625.
4. S. Shibata, I. Imaseki and M. Yamazaki: Chem.Pharm.Bull.(Tokyo), 7 , 449 (1959)
5. R.H.F. Manske and H.L. Holmes: The Alkaloids, Vol. III, pp 339 (Academic Press, New York and London, 1953)
6. E. Leete: Chem. & Ind., 1958 , 1088
7. M. Yamazaki , A. Ikuta , T. Mori and T. Tanaka: Tetrahedron Letters , 1967 , 3317
8. S. Akabori and K. Momotani: Proc.Imp. Acad. (Tokyo), 17 , 506 (1941).
9. R. McGrath, L.C. Vining, F. Sala and S. Westlake: Can.J.Biochem. , 46, 587 (1968)